

Tetrahedron: Asymmetry 10 (1999) 1335-1351

# Non-thiazolidinedione antihyperglycaemic agents. Part 4: Synthesis of $(\pm)$ -, (R)-(+)- and (S)-(-)-enantiomers of 2-oxy-3-arylpropanoic acids $^{\dagger}$

David Haigh,<sup>a,\*</sup> Helen C. Birrell,<sup>b</sup> Barrie C. C. Cantello,<sup>a</sup> Richard M. Hindley,<sup>a</sup> Anantha Ramaswamy,<sup>b</sup> Harshad K. Rami <sup>a</sup> and Nicola C. Stevens <sup>b</sup>

<sup>a</sup>Department of Medicinal Chemistry, SmithKline Beecham Pharmaceuticals, New Frontiers Science Park, The Pinnacles, Harlow, Essex, CM19 5AW, UK

Received 5 March 1999; accepted 18 March 1999

#### **Abstract**

The synthesis of a new series of potent 2-oxy-3-arylpropanoic acid antihyperglycaemic agents in both racemic and non-racemic form is described. Resolution of racemic acids  $\mathbf 1$  is accomplished via amide formation with either (S)-2-phenylglycinol or (S)-4-benzyloxazolidin-2-one as complementary resolving agents. © 1999 Elsevier Science Ltd. All rights reserved.

#### 1. Introduction

Non-insulin-dependent diabetes mellitus (NIDDM) is a complex metabolic disorder characterised by insulin resistance in the liver and peripheral tissues.<sup>2</sup> NIDDM affects up to 5% of the population of most western industrialised nations. Improvement of glycaemic control via reduction of peripheral tissue insulin resistance has long been recognised as pivotal to the development of an effective long-term therapy for the disease.<sup>3</sup> Thiazolidine-2,4-dione insulin sensitisers,<sup>4</sup> such as Rezulin (troglitazone)<sup>5</sup> and our own Phase-III clinical drug Avandia (rosiglitazone, BRL-49653),<sup>6</sup> are members of a new family of drugs used for the treatment of chronic NIDDM. Recent evidence suggests that compounds of this class exert their antidiabetic activity via activation of a nuclear hormone receptor PPARγ (peroxisomal proliferator-activated receptor γ).<sup>7</sup>

<sup>&</sup>lt;sup>b</sup>Department of Analytical Sciences, SmithKline Beecham Pharmaceuticals, New Frontiers Science Park, The Pinnacles, Harlow, Essex, CM19 5AW, UK

<sup>\*</sup> Corresponding author. Tel: 44-1279-627814; fax 44-1279-627841; e-mail: david\_haigh-1@sbphrd.com

<sup>&</sup>lt;sup>†</sup> For the previous paper in this series, see Haigh et al.<sup>1</sup>

BRL 48482

1 a, 
$$R^1 = Et$$
, d,  $R^1 = CH_2Ph$ ,
b,  $R^1 = CH_2CF_3$ , e,  $R^1 = m-C_6H_4CF_3$ ,
c,  $R^1 = CH_2CH_2OMe$ , f,  $R^1 = Me$ 

In the first paper of this series<sup>8</sup> we reported on the preliminary results of additional structure–activity studies on BRL 48482 (a compound equipotent to BRL 49653),<sup>6</sup> in which the 5-benzylthiazolidine-2,4-dione moiety of the former agent was replaced by a series of racemic 2-oxy-3-arylcarboxylic acids 1. In an initial examination of the enantioselectivity of PPARy binding and antidiabetic potency of these compounds,<sup>1</sup> we recently described an enzymic resolution approach to the synthesis of the (*R*)-(+)- and (*S*)-(-)-enantiomers of the 2-methoxy-acid 1f, together with an alternative preparation of the enantiomers of the 2-benzyloxy-analogues 1d. In order to undertake a more thorough examination of the impact of chirality in this area<sup>9</sup> we subsequently required larger quantities of both enantiomers of the key lead compounds for in vivo biological evaluation. This paper describes a new synthesis of racemic 2-alkoxy(aryloxy)-3-arylpropanoic acids 1a–e and the preparation of their enantiomers by two complementary resolution methodologies.

#### 2. Results and discussion

## 2.1. Synthesis of racemic 2-oxy-3-arylpropanoic acids

The racemic acids **1a**–**e**, chosen for the present resolution studies, were prepared according to Scheme 1. The Wadsworth–Emmons reaction of the appropriate alkoxy(aryloxy)phosphonoacetate **3** (prepared from triethyl diazophosphonoacetate and the alcohol in the presence of rhodium acetate)<sup>10,11</sup> with aldehyde **2**<sup>6</sup> afforded mixtures of geometric isomers of 2-oxy-3-arylpropenoates **4**. In agreement with an earlier report<sup>11a</sup> the alkoxy compounds **4a**–**d** were predominantly isolated as the Z-isomer (Table 1), although the ratio varied according to the substituent.<sup>12,13</sup> As expected,<sup>11b</sup> the *m*-trifluoromethylphenoxy-analogue **4e** was isolated as a 1:1 mixture of isomers. Reduction of **4** either by catalytic hydrogenation over palladium charcoal (for **4a**,**b**), or by dissolving magnesium metal in methanol solution<sup>14</sup> gave esters **5**. In the latter reaction, products **5c**–**e** were isolated as the methyl esters following in situ transesterification. Hydrolysis of **5** subsequently afforded the desired acids **1a**–**e** (Table 1).

#### 2.2. Chemical resolution of 2-oxy-3-arylpropanoic acids

Resolution of racemic α-functionalised carboxylic acids has classically been achieved via formation and fractional crystallisation of salts of a chiral amine.<sup>15</sup> However, in view of the unpredictability of the method and difficulties we had previously encountered in the choice of a suitable amine to resolve closely related thiazolidine-2,4-dione analogues,<sup>16</sup> this approach was deemed unattractive. Alternative approaches to the resolution of functionalised carboxylic acids have included kinetic<sup>17</sup> and dynamic kinetic<sup>18</sup> resolution and the formation and chromatographic separation<sup>19</sup> of diastereoisomeric esters by reaction of the acid with a suitable chiral alcohol.<sup>20</sup> Unfortunately, attempts to resolve racemic acids

Scheme 1. For definition of substituents R<sup>1</sup> and R<sup>2</sup> see Table 1. Reagents: (i) NaH, THF then **2**. (ii) Either (A) H<sub>2</sub>, 10% Pd–C, EtOH, or (B) Mg, MeOH. (iii) NaOH, MeOH

Substituent R1 Olefin 4 (±)-Acid 1  $(\pm)$ -Ester 5  $R^2$ Yield (%) E:Z ratio Methoda Yield (%) Yield (%) Et 73 a 71 38:62 A Et 50 b 97 CH<sub>2</sub>CF<sub>3</sub> 88 43:57 A Et 93 CH2CH2OMe 60 24:76 В Meb 37 92 d CH<sub>2</sub>Ph 68 41:59 В Meb 84 91

Table 1 Synthesis of racemic acids

50:50

94

m-C<sub>6</sub>H<sub>4</sub>CF<sub>3</sub>

1 via formation of diastereoisomeric esters of (S)-1-phenylethanol or of (R)-pantolactone<sup>20 d</sup> afforded chromatographically inseparable mixtures.

В

Meb

78

89

In an extension of this concept, the formation and separation of diastereoisomeric amides has also been used for the resolution of racemic acids<sup>21</sup> and Helmchen<sup>22</sup> has shown 2-phenylglycinol to be a particularly useful amine in this context. Reaction of the acid chlorides derived from  $1\mathbf{a}$ — $\mathbf{c}$  with (S)-2-phenylglycinol afforded amides which were easily separated by silica gel chromatography to give the individual[2R,N(1S)]-diastereoisomers  $6\mathbf{a}$ — $\mathbf{c}$  (faster eluting isomer) and [2S,N(1S)]-diastereoisomers  $7\mathbf{a}$ — $\mathbf{c}$  (slower eluting isomer), respectively, in excellent diastereoisomeric excess (often >98%; Scheme 2; Table 2). Subsequent acid hydrolysis afforded the corresponding (R)- and (S)-enantiomers  $8\mathbf{a}$ — $\mathbf{c}$  and  $9\mathbf{a}$ — $\mathbf{c}$ , respectively, in good enantiomeric excess (88–96%), although in only modest chemical yield (Table 2). However, the diastereoisomeric amides derived from acids  $1\mathbf{d}$  or  $1\mathbf{e}$  and either (S)-2-phenylglycinol or (S)-alaninol were inseparable by chromatography.

The use of various chiral oxazolidin-2-ones in the formation and separation of diastereoisomeric imide derivatives of racemic carboxylic acids has recently been reported.<sup>23</sup> However, to our knowledge, only a single example of the resolution of an  $\alpha$ -oxyacid in this manner has been recorded.<sup>23a</sup> Consequently, we decided to examine the suitability of the Evans chiral auxiliary (*S*)-4-benzyloxazolidin-2-one **10** for the resolution of the racemic acids **1d** and **1e**. Auxiliary **10** was acylated by the acid chlorides derived from

<sup>&</sup>lt;sup>a</sup>Reduction method (A): H<sub>2</sub>, 10% Pd-C, ethanol; (B): Magnesium, methanol.

bMethyl ester formed by *in-situ* transesterification.

Scheme 2. Resolution of  $(\pm)$ -1 via (S)-2-phenylglycinol. Reagents: (i)  $(COCl)_2$ , benzene. (ii) (S)-2-Phenylglycinol,  $NEt_3$ ,  $CH_2Cl_2$ . (iii)  $1 \text{ M } H_2SO_4$ , aq. dioxan

S	ubstituent R <sup>1</sup>	Diastereoisomeric Amidesa				Enantiomeric Acids				
		6		7		(R)-8		(S)-9		
		Yield	d.e.	Yield	d.e.	Yield	e.e.	Yield	e.e.	
		(%)b	(%)	(%) <sup>b</sup>	(%)	(%)	(%)	(%)	(%)	
a	Et	32	98.4	33	100	37	93.2	38	93.6	
b	CH <sub>2</sub> CF <sub>3</sub>	32	100	34	99	51	96.6	48	95.2	
c	CH <sub>2</sub> CH <sub>2</sub> OMe	44	90.8	27	92.6	60	94.0	43	88.0c	

Table 2 Resolution of  $(\pm)$ -1 via (S)-2-phenylglycinol

1d and 1e in a standard manner.<sup>24</sup> The resulting diastereoisomeric acyloxazolidinones 11d,e and 12d,e (faster and slower eluting isomers, respectively) were readily separable by chromatography (Scheme 3; Table 3). Subsequent cleavage of the auxiliary by treatment with sodium methoxide,<sup>25</sup> followed by acid hydrolysis of the resulting esters 13 and 14 afforded the desired enantiomeric acids 8d,e and 9d,e. It is of note that diastereoisomer separation was not possible when acids 1a and 1b were employed in this sequence; hence it would appear that (S)-2-phenylglycinol and (S)-4-benzyloxazolidin-2-one are complementary in their suitability for the resolution of 2-oxycarboxylic acids of this type.

### 2.3. Determination of absolute configuration

Following Helmchen's proposal of a conformational model for the determination of chromatographic elution order<sup>21 a</sup> and determination of absolute stereochemistry<sup>21 a,b,d</sup> of the pair of diastereoisomeric

<sup>&</sup>lt;sup>a</sup>Diastereoisomer **6** was the first-eluted isomer in each case.

bMaximum yield 50%.

<sup>&</sup>lt;sup>c</sup>E.e. improved to 94% by repeated crystallisation of the (S)-1-phenylethylamine salt.

Scheme 3. Resolution of  $(\pm)$ -1 via (S)-4-benzyloxazolidin-2-one. Reagents: (i) 1,  $(COCl)_2$ , benzene. (ii) 10, n-BuLi, THF, then add acid chloride. (iii) 1.1 equiv. NaOMe, MeOH. (iv) 2 M HCl, dioxan

Table 3 Resolution of  $(\pm)$ -1 via (S)-4-benzyloxazolidin-2-one

Substituent R <sup>1</sup>		Diastereoisomeric Imidesa				Enantiomeric Acids				
		11		12		(R)-8		(S)-9		
		Yield (%)b	d.e. (%)	Yield (%) <sup>b</sup>	d.e. (%)	Yield (%) <sup>c</sup>	e.e. (%)	Yield (%) <sup>c</sup>	e.e. (%)	
d	CH <sub>2</sub> Ph	24	97.8	17	89.8	55	96	54	88	
e	m-C <sub>6</sub> H <sub>4</sub> CF <sub>3</sub>	28	100	19	97.2	75	98	73	98	

<sup>&</sup>lt;sup>a</sup>Diastereoisomer 11 was the first-eluted isomer in each case.

amides formed by the reaction of a racemic carboxylic acid with a chiral amine (or vice versa), the generality of this model for predicting stereochemistry has been firmly established.<sup>26</sup> Thus, inspection of the conformational models and 'extended Newman projections' (arrowed) for (R,S)- and (S,S)-diastereoisomers **6** and **7**, Figs. 1 and 2, respectively, suggested that the methylene protons of the ArCH<sub>2</sub>–group in **7** should be shielded relative to **6** by the phenyl ring of the auxiliary.<sup>21a</sup> Comparison of the <sup>1</sup>H NMR spectra for **6a**–**c** and **7a**–**c** showed a consistent upfield shift of these protons in **7** ( $-0.05 \le \Delta \delta \le -0.15$  ppm). However, the protons directly attached to the stereogenic centre of the acid moiety showed no <sup>1</sup>H NMR diastereoisomeric shifts. Figs. 1 and 2 also predicted that the (R,S)-diastereoisomer **6** should be the chromatographically earlier eluting isomer, since the bulky phenyl and aryl groups occupy opposite sides of the plane of the amide group.<sup>21a</sup> In each case, diastereoisomer **6** was indeed the chromatographically earlier eluting isomer. Together, these data thus supported the assignment of the stereochemistry of **6** 

bMax. yield 50%.

<sup>&</sup>lt;sup>c</sup>Overall yield from imide

$$\stackrel{\mathsf{ArCH}_2}{=} \stackrel{\mathsf{O}}{=} \stackrel{\mathsf{H}}{=} \stackrel{\mathsf{O}}{=} \stackrel{\mathsf{H}}{=} \stackrel{\mathsf{OH}}{=} \stackrel{\mathsf{ArCH}_2}{=} \stackrel{\mathsf{H}}{=} \stackrel{\mathsf{H}}{$$

Figure 1. (R,S)-Diastereoisomer 6

$$ArCH_2 \xrightarrow{O} OR \xrightarrow{H} OH = HOCH_2 OR$$

Figure 2. (S,S)-Diastereoisomer 7

and **7** and hence of the corresponding enantiomeric acids **8a–c** and **9a–c**. Independent proof of absolute stereochemistry was subsequently provided by correlation of **8a–c** and **9a–c** with the enantiomers of **1d** and **1f**, whose stereochemistry had been unambiguously determined by X-ray crystallography.<sup>1</sup>

Similar observations were also noted with the diastereoisomers 11d,e and 12d,e. In each case, (R,S)-diastereoisomer 11 was the faster eluting. The protons attached to the two stereogenic centres in 11 and 12 were also found to show diagnostic shift differences in the  $^1H$  NMR spectra. Diastereoisomer 11 consistently showed an upfield shift for the proton on the stereogenic centre in the acid moiety $^{\ddagger}$  and a downfield shift for the proton on the stereogenic centre in the auxiliary, relative to 12.

#### 3. Conclusion

Synthesis of a new series of potent, racemic, 2-oxy-3-arylpropanoic acid antihyperglycaemic agents 1 has been reported. Resolution of acids 1 was achieved by the complementary utilisation of either (*S*)-2-phenylglycinol or (*S*)-4-benzyloxazolidin-2-one as resolving agents. Use of the latter auxiliary represents an extension to the versatility of this oxazolidin-2-one in synthesis. The absolute stereochemistry of acids 8 and 9 was inferred by literature analogy based on chromatographic elution order and <sup>1</sup>H NMR chemical shift differences between the diastereoisomeric amide intermediates and was substantiated by correlation to earlier X-ray studies.

Biological evaluation of the enantiomeric acids **8** and **9** demonstrated the greater potency of the (*S*)-enantiomer **9** and will be reported elsewhere. However, a de novo enantioselective aldol synthesis of **9a**–**e** is the subject of the following paper.<sup>27</sup>

#### 4. Experimental

#### 4.1. General experimental details

Mass spectroscopy was conducted using electron impact (EI), chemical ionisation (CI), with ammonia as the reagent gas, or fast atom bombardment (FAB) in a 3-nitrobenzyl alcohol–sodium acetate (NOBA–Na) matrix. Compounds characterised by high resolution mass measurement were homogeneous by TLC. <sup>1</sup>H NMR spectra were recorded at 270 or 400 MHz in CDCl<sub>3</sub> solution. Chemical shifts are given

<sup>&</sup>lt;sup>‡</sup> The <sup>1</sup>H NMR signal for the proton on the stereogenic centre in the acid moiety is particularly diagnostic.  $\Delta\delta$  ( $\delta_{11}$ – $\delta_{12}$ ) for CHOR<sup>1</sup> is –0.12 ppm and –0.16 ppm for **11d/12d** (R<sup>1</sup>=CH<sub>2</sub>Ph) and **11e/12e** (R<sup>1</sup>=m-C<sub>6</sub>H<sub>4</sub>CF<sub>3</sub>), respectively. Similar shifts were observed for the (inseparable) diastereoisomers corresponding to **11a/12a** (R<sup>1</sup>=Et,  $\Delta\delta$ =–0.14 ppm) and **11b/12b** (R<sup>1</sup>=CH<sub>2</sub>CF<sub>3</sub>,  $\Delta\delta$ =–0.12 ppm).

in  $\delta$  (ppm) relative to TMS and coupling constants J are given in hertz.  $[\alpha]_D^{25}$  values are given in deg cm<sup>2</sup> g<sup>-1</sup>. Dry solvents refer to the use of Aldrich Sure-Seal<sup>TM</sup> dried solvents. All organic solutions obtained from aqueous extractions were dried over MgSO<sub>4</sub>. Chromatography refers to flash chromatography on silica gel.

#### 4.2. HPLC conditions

Diastereoisomeric amides **6a,c** and **7a,c** were resolved on a Zorbax Octyl column using MeOH:0.05 M NH<sub>4</sub>OAc (pH 4.5, 70:30 v/v) as eluent. Amides **6b** and **7b** were resolved on the same column using MeOH:0.1 M NH<sub>4</sub>OAc (pH 4.5, 65:35 v/v). Diastereoisomeric imides **11** and **12** were separated on a Lichrosorb Si60 column using hexane:ethyl acetate (70:30 v/v) as eluent. Chiral esters **13d** and **14d** were resolved on a Chiralcel OD column using hexane:Pr<sup>i</sup>OH (95:5 v/v) and for esters **13e** and **14e**, a Chiralpak AD column using hexane:EtOH (90:10 v/v) as eluent was employed. Chiral acids **8a,d** and **9a,d** were separated on a Chiralpak AD column using hexane:Pr<sup>i</sup>OH (92:8 v/v) containing CF<sub>3</sub>CO<sub>2</sub>H (0.05% v/v) as eluent; for acids **8b** and **9b**, hexane:EtOH (96.5:3.5 v/v) containing CF<sub>3</sub>CO<sub>2</sub>H (0.05% v/v) was used and for acids **8c** and **9c**, hexane:EtOH (90:10 v/v) containing CF<sub>3</sub>CO<sub>2</sub>H (0.05% v/v) was used. Acids **8e** and **9e** were resolved on a Chiralcel OJ column using hexane:EtOH (85:15 v/v) containing CF<sub>3</sub>CO<sub>2</sub>H (0.05% v/v) as eluent. Solvent flow rate was 1 mL min<sup>-1</sup> throughout and detection wavelength was either 220, 245 or 250 nm.

## 4.3. General procedure for preparation of 2-alkoxy(aryloxy)phosphonoacetates 3

### 4.3.1. Ethyl 2-(2,2,2-trifluoroethoxy)-2-diethylphosphonoacetate 3b

A mixture of ethyl 2-diazo-2-diethylphosphonoacetate (3.75 g, 0.015 mol), 2,2,2-trifluoroethanol (2.19 mL, 0.03 mol) and rhodium(II) acetate dimer (66 mg, 1 mol%) in benzene (40 mL) was heated at reflux for 18 h. After removal of solvent under reduced pressure, the residual oil was chromatographed with ethyl acetate:diethyl ether (2.5:97.5) as eluent to give the product as a clear oil (3.66 g, 76%); (found M<sup>+</sup> (EI) 322.0795.  $C_{10}H_{18}F_{3}PO_{6}$  requires M 322.0793);  $v_{max}$  (film)/cm<sup>-1</sup> 1745 (CO) and 1260 [PO(OEt)<sub>2</sub>];  $\delta_{H}$  (270 MHz, CDCl<sub>3</sub>) 1.30–1.40 (9H, m, 3×CH<sub>3</sub>), 3.95 (1H, m, OCHHCF<sub>3</sub>), 4.10–4.40 (7H, m, OCHHCF<sub>3</sub> and 3×CH<sub>2</sub>) and 4.51 (1H, d,  $^{2}J_{PH}$  18.7, POCH); m/z (CI) 340 [(M+NH<sub>4</sub>)<sup>+</sup>, 100%], 323 [(M+H)<sup>+</sup>, 12] and 258 (9).

## 4.3.2. Ethyl 2-(2-methoxyethoxy)-2-diethylphosphonoacetate 3c

Similarly prepared. An oil, 66%; (found M<sup>+</sup> (EI) 298.1181.  $C_{11}H_{23}PO_7$  requires M 298.1181);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 1.32 (9H, m, 3×CH<sub>2</sub>CH<sub>3</sub>), 3.35 (3H, s, OMe), 3.59 (2H, t, J 4.5, CH<sub>2</sub>OMe), 3.80 (2H, m, OCH<sub>2</sub>CH<sub>2</sub>OMe), 4.20 (6H, m, 3×CH<sub>2</sub>CH<sub>3</sub>) and 4.47 (1H, d,  $^2J_{PH}$  18.9, POCH).

### 4.3.3. Ethyl 2-benzyloxy-2-diethylphosphonoacetate 3d

Similarly prepared. An oil, 56%; (found (M+H)<sup>+</sup> (CI) 331.1308.  $C_{15}H_{23}PO_6$  requires M 331.1311);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 1.30 (9H, m, 3×OCH<sub>2</sub>CH<sub>3</sub>), 4.20 (6H, 3×OCH<sub>2</sub>CH<sub>3</sub>), 4.35 (1H, d,  $^2J_{PH}$  18.9, POCH), 4.58 (1H, d, J 11.8, OCHHPh), 4.82 (1H, d, J 11.8, OCHHPh) and 7.28 (5H, m, Ph).

#### 4.4. General procedure for olefination of aldehyde 2

4.4.1. Ethyl (E/Z)-3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2,2,2-trifluoroethoxy)propenoate  ${\it 4b}$ 

Sodium hydride (60% dispersion in oil, 250 mg,  $6.2 \times 10^{-3}$  mol, 1.1 equiv.) was added to a solution of phosphonate **3b** (1.819 g,  $5.6 \times 10^{-3}$  mol) in dry THF (5 mL) at 0°C under an atmosphere of argon. After stirring at 0°C for 10 min, a solution of aldehyde **2** (1.670 g, 1 equiv.) in dry THF (5 mL) was added and the mixture allowed to warm to ambient temperature and stirred for 3.5 h. Water (30 mL) was added, the mixture was extracted with ethyl acetate (×3) and the combined extracts washed with brine then dried and evaporated to dryness to afford an oil. This was chromatographed using ethyl acetate:hexane (gradient, 10:90 to 50:50) as eluent to afford a 43:57 *E:Z* mixture of geometric isomers of the title compound as a wax (2.278 g, 88%); (found (M+H)<sup>+</sup> (FAB) 465. C<sub>23</sub>H<sub>23</sub>F<sub>3</sub>N<sub>2</sub>O<sub>5</sub> requires *M* 465);  $\nu_{\text{max}}$  (film)/cm<sup>-1</sup> 1712 (CO);  $\delta_{\text{H}}$  (270 MHz, CDCl<sub>3</sub>) geometric isomerism causes doubling of most peaks, 1.17, 1.36 (combined, 3H, t, *J* 7.1, isomers of OCH<sub>2</sub>CH<sub>3</sub>), 3.35, 3.36 (combined, 3H, s, NMe), 3.95 (2H, m, OCH<sub>2</sub>CH<sub>2</sub>N), 4.10–4.40 (6H, m, 3×OCH<sub>2</sub>), 6.48 (0.43H, s, *E*-olefinic H) and 6.80–7.70 (8.57H, m, *Z*-olefinic H and ArH); m/z (FAB) 487 [(M+Na)<sup>+</sup>, 10%], 465 [(M+H)<sup>+</sup>, 100], 419 (4), 316 (16) and 175 (61).

4.4.2. Ethyl (E/Z)-3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-ethoxypropenoate 4a Similarly prepared. A gum, 71%; 38:62 E:Z mixture of geometric isomers; (found M<sup>+</sup> (EI) 410.1840. C<sub>23</sub>H<sub>26</sub>N<sub>2</sub>O<sub>5</sub> requires M 410.1842); δ<sub>H</sub> (270 MHz, CDCl<sub>3</sub>) geometric isomerism causes doubling of most peaks, 1.13 (0.38×3H, t, J 7.1, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, E-isomer), 1.25–1.45 (3H+0.62×3H, m, OCH<sub>2</sub>CH<sub>3</sub> and CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>, Z-isomer), 3.34 (3H, s, NMe), 3.95–4.40 (8H, m, 3×OCH<sub>2</sub> and NCH<sub>2</sub>), 6.06 (0.32H, s, E-olefinic H) and 6.80–7.80 (8.62H, m, Z-olefinic H and aryl-H).

## 4.4.3. Ethyl (E/Z)-3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2-methoxyethoxy)-propenoate $\mathbf{4c}$

Similarly prepared. A gum, 60%; 24:76 E:Z mixture of geometric isomers; (found M<sup>+</sup> (EI) 440.1948.  $C_{24}H_{28}N_2O_6$  requires M 440.1948);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) geometric isomerism causes doubling of most peaks, 1.12, 1.35 (combined, 3H, t, J 7.1, isomers of OCH<sub>2</sub>CH<sub>3</sub>), 3.35–3.45 (combined, 6H, m, isomeric NMe and OMe), 3.67, 3.72 (combined, 2H, m, OCH<sub>2</sub>CH<sub>2</sub>OMe), 3.90–4.35 (8H, m, 3×OCH<sub>2</sub> and NCH<sub>2</sub>), 6.15 (0.24H, s, E-olefinic H) and 6.80–7.80 (8.76H, m, E-olefinic H and ArH).

## 4.4.4. Ethyl (E/Z)-3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-benzyloxypropenoate **4d**

Similarly prepared. A gum, 68%; 41:59 E:Z mixture of geometric isomers. These were partially resolved by silica gel chromatography using ethyl acetate:dichloromethane (2:98) as an eluent to afford initially the Z-isomer Z-4d: (found M<sup>+</sup> (EI) 472.1996.  $C_{28}H_{28}N_2O_5$  requires M 472.1996);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 1.34 (3H, t, J 7.1, OCH<sub>2</sub>CH<sub>3</sub>), 3.34 (3H, s, NMe), 3.95 (2H, t, J 5.2, NCH<sub>2</sub>), 4.30 (4H, m, 2×OCH<sub>2</sub>), 4.93 (2H, s, CH<sub>2</sub>Ph), 6.83 (2H, d, J 8.9, aryl 3-H and 5-H) and 6.95–7.75 (12H, m, aryl-H and Z-olefinic H); followed by the E-isomer E-4d:  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 1.15 (3H, t, J 7.2, OCH<sub>2</sub>CH<sub>3</sub>), 3.34 (3H, s, NMe), 3.94 (2H, t, J 5.3, NCH<sub>2</sub>), 4.16 (2H, q, J 7.2 OCH<sub>2</sub>CH<sub>3</sub>), 4.25 (2H, t, J 5.3 OCH<sub>2</sub>), 4.94 (2H, s, CH<sub>2</sub>Ph), 6.19 (1H, s, E-olefinic H) and 6.78–7.45 (13H, m, aryl-H).

4.4.5. Ethyl (E/Z)-3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(3-trifluoromethyl-phenoxy)propenoate  $\mathbf{4e}$ 

Similarly prepared. A gum, 94%; 50:50 E:Z mixture of geometric isomers; (found M<sup>+</sup> (EI) 526.1712.  $C_{28}H_{25}N_2O_5$  requires M 526.1716);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) geometric isomerism causes doubling of most peaks, 1.05, 1.18 (combined, 3H, t, J 7.1, isomers of OCH<sub>2</sub>CH<sub>3</sub>), 3.32, 3.35 (combined, 3H, s, NMe), 3.96 (2H, m, NCH<sub>2</sub>), 4.05–4.35 (4H, m, 2×OCH<sub>2</sub>) and 6.80–7.70 (13H, m, aryl-H and olefinic H).

## 4.5. Reduction of alkene 4, Method A

4.5.1. Ethyl 3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2,2,2-trifluoroethoxy)-propanoate  $\mathbf{5b}$  ( $R^2 = Et$ )

A mixture of alkene **4b** (2.240 g,  $4.83 \times 10^{-3}$  mol), 10% palladium on charcoal (220 mg) and ethanol (30 mL) was shaken under an atmosphere of hydrogen at 30 p.s.i. and ambient temperature. After 5 h, the catalyst was removed by filtration and solvent removed in vacuo to give the product as an oil (2.201 g, 98%), which was used without further purification; (found M<sup>+</sup> (EI) 466.1715.  $C_{23}H_{25}F_3N_2O_5$  requires M 466.1716);  $\nu_{max}$  (film)/cm<sup>-1</sup> 1744 (CO);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 1.24 (3H, t, J 7.2, OCH<sub>2</sub>CH<sub>3</sub>), 3.00 (2H, m, ArCH<sub>2</sub>CH), 3.34 (3H, s, NMe), 3.65 (1H, m, OCHHCF<sub>3</sub>), 3.94 (2H, t, J 5.2, NCH<sub>2</sub>), 3.97 (1H, m, OCHHCF<sub>3</sub>), 4.10–4.30 (5H, m, 2×OCH<sub>2</sub> and ArCH<sub>2</sub>CH), 6.80 (2H, d, J 8.8, phenyl-3H and 5H) and 6.95–7.40 (6H, m, ArH); m/z (FAB) 489 [(M+Na)<sup>+</sup>, 17%], 467 [(M+H)<sup>+</sup>, 100], 175 (52) 161 (18) and 148 (58).

4.5.2. Ethyl 3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-ethoxypropanoate 5a ( $R^2$ =Et)

Similarly prepared. A gum, 73%; (found (M+H)<sup>+</sup> (CI) 413.2070.  $C_{23}H_{28}N_2O_5$  requires M 413.2077);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 1.15 (3H, t, J 6.9, OCH<sub>2</sub>CH<sub>3</sub>), 1.22 (3H, t, J 7.1, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 2.95 (2H, m, ArCH<sub>2</sub>CH), 3.30 (1H, m, OCHHCH<sub>3</sub>), 3.34 (3H, s, NMe), 3.60 (1H, m, OCHHCH<sub>3</sub>), 3.93 (3H, m, ArCH<sub>2</sub>CH and NCH<sub>2</sub>), 4.17 (2H, q, J 7.1, CO<sub>2</sub>CH<sub>2</sub>CH<sub>3</sub>), 4.24 (2H, t, J 5.2, OCH<sub>2</sub>), 6.80 (2H, d, J 8.8, aryl 3-H and 5-H) and 6.95–7.40 (6H, m, aryl H).

### 4.6. Reduction of alkene 4, Method B

4.6.1. Methyl 3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2-methoxyethoxy)-propanoate  $\mathbf{5c}$  ( $R^2$ =Me)

A mixture of alkene **4c** (1.36 g, 3 mmol), magnesium (0.5 g, 20 mmol) iodine (ca. 10 mg) and methanol (80 mL) was warmed gently (hairdryer) with stirring until evolution of hydrogen commenced. An additional portion of magnesium (3.5 g, 140 mmol) was added and the mixture stirred at room temperature (temperature regulated by immersion in a cold water bath) for 23 h then concentrated. The residue was suspended in water (500 mL) and concentrated hydrochloric acid was added to give a final pH of 2, once all the solid had dissolved. The mixture was extracted with ethyl acetate (3×500 mL) and the combined organic solutions washed with brine (500 mL), dried and evaporated. The residue was chromatographed using ethyl acetate:dichloromethane (gradient, 5:95 to 10:90) as eluent to afford 5c ( $R^2$ =Me), a colourless gum, 0.476 g, 37%; (found (M+H)<sup>+</sup> (CI) 428.1946.  $C_{23}H_{28}N_2O_6$  requires M 428.1948);  $V_{max}$  (film)/cm<sup>-1</sup> 1748 (CO);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 2.95 (2H, m, ArCH<sub>2</sub>CH), 3.29 (3H, s, OMe), 3.35 (3H, s, NMe), 3.47 (3H, m, OCHHCH<sub>2</sub>OMe), 3.69 (4H, m, CO<sub>2</sub>Me and OCHHCH<sub>2</sub>OMe),

3.93 (2H, t, J 5.2, NCH<sub>2</sub>), 4.06 (1H, t, J 5.8, ArCH<sub>2</sub>CH), 4.24 (2H, t, J 5.2, OCH<sub>2</sub>), 6.80 (2H, d, J 8.8, aryl 3-H and 5-H) and 6.95–7.40 (6H, m, aryl H); m/z (CI) 429 [(M+H)<sup>+</sup>, 100%], 175 (15) and 148 (10).

4.6.2. Methyl 3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-benzyloxypropanoate**5d** $(<math>R^2=Me$ )

Similarly prepared. A gum, 84%; (found M<sup>+</sup> (EI) 460.2000.  $C_{27}H_{28}N_2O_5$  requires M 460.1999);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 2.98 (2H, m, ArCH<sub>2</sub>CH), 3.35 (3H, s, NMe), 3.69 (3H, s, OMe), 3.94 (2H, t, J 5.2, NCH<sub>2</sub>), 4.10 (1H, dd, J 7.7 and 5.5, ArCH<sub>2</sub>CH), 4.24 (2H, t, J 5.2, OCH<sub>2</sub>), 4.35 (1H, d, J 11.9, OCHHPh), 4.64 (1H, d, J 11.9, OCHHPh), 6.79 (2H, d, J 8.8, aryl 3-H and 5-H) and 6.95–7.40 (11H, m, aryl-H).

4.6.3. Methyl 3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(3-trifluoromethylphenoxy)propanoate **5e** 

Similarly prepared. (R<sup>2</sup>=Me), a gum, 78%; (found M<sup>+</sup> (EI) 514.1717.  $C_{27}H_{25}F_3N_2O_5$  requires M 514.1716);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 3.18 (2H, d, J 6.4, ArC $H_2$ CH), 3.33 (3H, s, NMe), 3.71 (3H, s, OMe), 3.93 (2H, t, J 5.2, NCH<sub>2</sub>), 4.23 (2H, t, J 5.2, OCH<sub>2</sub>), 4.77 (1H, t, J 6.4, ArC $H_2$ CH), 6.83 (2H, d, J 8.8, aryl 3-H and 5-H) and 6.90–7.35 (10H, m, aryl-H).

### 4.7. Alkaline hydrolysis of esters 5

4.7.1.  $(\pm)$ -3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-ethoxypropanoic acid 1a

Aqueous sodium hydroxide hydrolysis of  $\bf 5a$  afforded acid  $\bf 1a$  as a white solid (50%), m.p. 109–110°C; (found C, 65.6; H, 6.3; N, 7.4%; M<sup>+</sup> (EI) 384.1686. C<sub>21</sub>H<sub>24</sub>N<sub>2</sub>O<sub>5</sub> requires C, 65.6; H, 6.3; N, 7.3%; *M* 384.1685);  $v_{max}$  (KBr)/cm<sup>-1</sup> 3000–2500 (COOH), 1710 (CO);  $\delta_{H}$  (270 MHz, CDCl<sub>3</sub>) 1.18 (3H, t, *J* 6.9, OCH<sub>2</sub>CH<sub>3</sub>), 2.95 (1H, dd, *J* 14.0 and 7.2, ArCHHCH), 3.05 (1H, dd, *J* 14.0 and 5.0, ArCHHCH), 3.32 (3H, s, NMe), 3.45 (1H, m, OCHHCH<sub>3</sub>), 3.60 (1H, m, OCHHCH<sub>3</sub>), 3.91 (2H, t, *J* 5.0, NCH<sub>2</sub>), 4.04 (1H, dd, *J* 7.2 and 5.0, ArCH<sub>2</sub>CH), 4.18 (2H, t, *J* 5.0, OCH<sub>2</sub>), 5.00 (1H, br, exchanges with D<sub>2</sub>O, CO<sub>2</sub>H), 6.79 (2H, d, *J* 8.8, aryl 3-H and 5-H) and 6.95–7.40 (6H, m, aryl-H); m/z (EI) 384 (M<sup>+</sup>, 9%), 311 (40), 175 (100), 161 (72) and 148 (100).

4.7.2.  $(\pm)$ -3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2,2,2-trifluoroethoxy)-propanoic acid  ${\it 1b}$ 

Similarly prepared. A solid, 93%; m.p.  $116-117^{\circ}$ C; (found C, 57.4; H, 4.9; N, 6.4%; M<sup>+</sup> (EI) 438.1403. C<sub>21</sub>H<sub>21</sub>F<sub>3</sub>N<sub>2</sub>O<sub>5</sub> requires C, 57.5; H, 4.8; N, 6.4%; *M* 438.1403);  $\delta_{\rm H}$  (270 MHz, CDCl<sub>3</sub>) 3.11 (2H, m, ArCH<sub>2</sub>CH), 3.29 (3H, s, NMe), 3.76 (1H, m, OCHHCF<sub>3</sub>), 3.85 (2H, m, NCH<sub>2</sub>), 4.02 (2H, m, OCH<sub>2</sub>CH<sub>2</sub>N), 4.10 (1H, m, OCHHCF<sub>3</sub>), 4.24 (1H, m, ArCH<sub>2</sub>CH), 6.75 (2H, d, *J* 8.8, phenyl 3-H and 5-H), 6.95–7.40 (6H, m, ArH) and 8.75 (1H, br, exchanges with D<sub>2</sub>O, COOH).

4.7.3. ( $\pm$ )-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2-methoxyethoxy)-propanoic acid  $\bf 1c$ 

Similarly prepared. A solid, 92%; m.p. 87–89°C; (found C, 63.8; H, 6.5; N, 7.05%; M<sup>+</sup> (EI) 414.1791.  $C_{21}H_{26}N_2O_6$  requires C, 63.8; H, 6.3; N, 6.8%; *M* 414.1791);  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 2.91 (1H, dd, *J* 14.3 and 8.5, ArCHHCH), 3.14 (1H, dd, *J* 14.3 and 3.8, ArCHHCH), 3.33 (3H, s, NMe), 3.38 (3H, s, OMe), 3.40–3.70 (5H, m, reduces to 4H on exchange with D<sub>2</sub>O, COOH and OCH<sub>2</sub>CH<sub>2</sub>OMe), 3.93 (2H, t, *J* 5.2, NCH<sub>2</sub>), 4.05 (1H, dd, *J* 8.5 and 3.8, ArCH<sub>2</sub>CH), 4.21 (2H, t, *J* 5.2, OCH<sub>2</sub>), 6.80 (2H, d, *J* 8.6, aryl 3-H and 5-H) and 6.96–7.40 (6H, m, aryl-H).

4.7.4. (±)-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-benzyloxypropanoic acid **1d** Similarly prepared. A foam, 91%; (found M<sup>+</sup> (EI) 446.1843. C<sub>26</sub>H<sub>26</sub>N<sub>2</sub>O<sub>5</sub> requires *M* 446.1842); δ<sub>H</sub> (270 MHz, CDCl<sub>3</sub>) 3.05 (2H, m, ArCH<sub>2</sub>CH), 3.31 (3H, s, NMe), 3.90 (2H, t, *J* 5.4, NCH<sub>2</sub>), 4.15 (3H, m, ArCH<sub>2</sub>CH and OCH<sub>2</sub>CH<sub>2</sub>N), 4.45 (1H, d, *J* 11.5, PhCHHO), 4.70 (1H, d, *J* 11.5, PhCHHO), 6.70–7.40 (13H, m, aryl-H) and 7.50 (1H, br, exchanges with D<sub>2</sub>O, COOH).

4.7.5.  $(\pm)$ -3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(3-trifluoromethylphenoxy)-propanoic acid  $\mathbf{1e}$ 

Similarly prepared. A solid, 89%; m.p.  $168-169^{\circ}$ C; (found C, 62.4; H, 4.6; N, 5.7%; M<sup>+</sup> (EI) 428.1946. C<sub>26</sub>H<sub>23</sub>F<sub>3</sub>N<sub>2</sub>O<sub>5</sub> requires C, 62.4; H, 4.6; N, 5.6%; *M* 428.1948);  $\delta_{\rm H}$  (270 MHz, CDCl<sub>3</sub>) 3.25 (5H, m, ArCH<sub>2</sub> and NMe), 3.83 (2H, t, *J* 4.9, NCH<sub>2</sub>), 4.00 (1H, br, exchanges with D2O, COOH), 4.04 (2H, t, *J* 4.9, OCH<sub>2</sub>), 4.86 (1H, t, *J* 6.0, ArCH<sub>2</sub>CH), 6.77 (2H, d, *J* 8.5, aryl 3-H and 5-H) and 6.95–7.40 (10H, m, aryl-H).

4.8. General procedure for reaction of racemic 1a–c with (S)-2-phenylglycinol. Preparation of diastereoisomeric  $\alpha$ -ethoxyamides 6a and 7a

Oxalyl chloride (3.4 mL, 39 mmol) was added dropwise to a solution of ethoxyacid 1a (2.96 g, 7.7 mmol) in dry benzene (100 mL). The resulting mixture was heated at reflux for 2 h, cooled and evaporated in vacuo to afford the acid chloride. This was dissolved in dry dichloromethane (150 mL) and the solution added, over 5 min, to an ice-cooled solution of (S)-2-phenylglycinol (1.056 g, 7.7 mmol) and triethylamine (2.2 mL, 15.4 mmol) in dichloromethane (150 mL) under argon. The mixture was allowed to warm to ambient temperature and stirred for an additional 16 h before being evaporated. The residue was chromatographed using isohexane:ethyl acetate (gradient, 25:75 to 0:100) as eluent to afford firstly the [2R,N(1S)]-diastereoisomeric amide 6a, followed by the [2S,N(1S)]-diastereoisomer 7a.

4.8.1. [2R,N(1S)]-3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-ethoxy-N-(2-hydroxy-1-phenylethyl)propanamide **6a** 

A colourless gum (1.222 g, 32%); (found (M+H)<sup>+</sup> (CI) 504.2491.  $C_{29}H_{33}N_{3}O_{5}$  requires M 504.2499);  $[\alpha]_{D}^{25}$  +21 (c 1.15 in CHCl<sub>3</sub>); d.e. 98.4% by HPLC;  $\nu_{max}$  (film)/cm<sup>-1</sup> 3420 (NH) and 1645 (CO);  $\delta_{H}$  (270 MHz, CDCl<sub>3</sub>) 1.12 (3H, t, J 7.0, OCH<sub>2</sub>Me), 2.65 (1H, br, exchanges with D<sub>2</sub>O, OH), 2.94 (1H, dd, J 14.1 and 6.3, ArCHHCH), 3.10 (1H, dd, J 14.1 and 3.8, ArCHHCH), 3.33 (3H, s, NMe), 3.50 (2H, m, OCH<sub>2</sub>Me), 3.65 (2H, m, CH<sub>2</sub>OH), 3.93 (2H, t, J 5.3, NCH<sub>2</sub>), 3.95 (1H, dd, J 6.3 and 3.8, ArCH<sub>2</sub>CH), 4.25 (2H, t, J 5.3, OCH<sub>2</sub>CH<sub>2</sub>N), 4.95 (1H, m, PhCHN) and 6.80–7.40 (14H, m, reduces to 13H on exchange with D<sub>2</sub>O, aryl-H and NH); m/z (CI) 504 [(M+H)<sup>+</sup>, 100%], 486 (5), 440 (10) and 412 (3).

4.8.2. [2S,N(1S)]-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-ethoxy-N-(2-hydroxy-1-phenylethyl)propanamide **7a** 

A colourless gum (1.278 g, 33%); (found (M+H)<sup>+</sup> (CI) 504.2486.  $C_{29}H_{33}N_{3}O_{5}$  requires M 504.2499);  $[\alpha]_{D}^{25}$  –19 (c 1.25 in CHCl<sub>3</sub>); d.e. 100% by HPLC;  $\nu_{max}$  (film)/cm<sup>-1</sup> 3420 (NH) and 1645 (CO);  $\delta_{H}$  (270 MHz, CDCl<sub>3</sub>) 1.16 (3H, t, J 7.0, OCH<sub>2</sub>Me), 2.60 (1H, br, exchanges with D<sub>2</sub>O, OH), 2.86 (1H, dd, J 14.2 and 6.8, ArCHHCH), 3.05 (1H, dd, J 14.2 and 3.7, ArCHHCH), 3.35 (3H, s, NMe), 3.55 (2H, m, OCH<sub>2</sub>Me), 3.83 (2H, m, CH<sub>2</sub>OH), 3.95 (3H, m, NCH<sub>2</sub> and ArCH<sub>2</sub>CH), 4.21 (2H, t, J 5.2, OCH<sub>2</sub>CH<sub>2</sub>N), 4.97 (1H, m, PhCHN) and 6.80–7.40 (14H, m, reduces to 13H on exchange with D<sub>2</sub>O, aryl-H and NH); m/z (CI) 504 [(M+H)<sup>+</sup>, 100%] and 486 (5).

4.8.3. [2R,N(1S)]-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2,2,2-trifluoro-ethoxy)-N-(2-hydroxy-1-phenylethyl)propanamide **6b** 

Similarly prepared. A foam, 32%; (found M<sup>+</sup> (EI) 557.2136.  $C_{29}H_{30}F_3N_3O_5$  requires M 557.2138); [ $\alpha$ ]<sub>D</sub><sup>25</sup> +39 (c 0.35 in MeOH); d.e. 100% by HPLC;  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 2.40 (1H, br, exchanges with D<sub>2</sub>O, OH), 3.00 (1H, dd, J 13.5 and 6.9, ArCHHCH), 3.18 (1H, dd, J 13.5 and 3.6, ArCHHCH), 3.33 (3H, s, NMe), 3.60–3.85 (4H, m, OCH<sub>2</sub>CF<sub>3</sub> and CH<sub>2</sub>OH), 3.94 (2H, t, J 5.2, NCH<sub>2</sub>CH<sub>2</sub>O), 4.11 (1H, dd, J 6.9 and 3.6, ArCH<sub>2</sub>CH), 4.26 (2H, t, J 5.2, NCH<sub>2</sub>CH<sub>2</sub>O), 5.00 (1H, m, PhCH), 6.83 (2H, d, J 8.8, aryl 3-H and 5-H) and 6.90–7.40 (12H, m, aryl-H and NH).

4.8.4. [2S,N(1S)]-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2,2,2-trifluoro-ethoxy)-N-(2-hydroxy-1-phenylethyl)propanamide **7b** 

Similarly prepared. A foam, 34%; (found M<sup>+</sup> (EI) 557.2136.  $C_{29}H_{30}F_3N_3O_5$  requires M 557.2138);  $[\alpha]_D^{25}$  +14 (c 0.50 in MeOH); d.e. 99% by HPLC;  $\delta_H$  (400 MHz, CDCl<sub>3</sub>) 2.30 (1H, br, exchanges with D<sub>2</sub>O, OH), 2.90 (1H, dd, J 14.4 and 7.3, ArCHHCH), 3.13 (1H, dd, J 14.4 and 3.6, ArCHHCH), 3.36 (3H, s, NMe), 3.70–3.87 (2H, m, OCH<sub>2</sub>CF<sub>3</sub>), 3.84 (2H, d, J 5.0, CH<sub>2</sub>OH), 3.95 (2H, t, J 5.2, NCH<sub>2</sub>CH<sub>2</sub>O), 4.12 (1H, dd, J 7.3 and 3.6, ArCH<sub>2</sub>CH), 4.22 (2H, t, J 5.2, NCH<sub>2</sub>CH<sub>2</sub>O), 5.01 (1H, m, PhCH), 6.75 (2H, d, J 8.7, aryl 3-H and 5-H) and 6.95–7.40 (12H, m, aryl-H and NH).

4.8.5.~[2R,N(1S)]-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2-methoxyethoxy)-N-(2-hydroxy-1-phenylethyl)propanamide  $\mathbf{6c}$ 

Similarly prepared. A gum, 44%; (found M<sup>+</sup> (EI) 533.2526.  $C_{30}H_{35}N_3O_6$  requires M 533.2526);  $[\alpha]_D^{25}$  +34 (c 1.7 in CHCl<sub>3</sub>); d.e. 90.8% by HPLC;  $\delta_H$  (400 MHz, CDCl<sub>3</sub>) 2.86 (1H, br, exchanges with D<sub>2</sub>O, OH), 2.92 (1H, dd, J 14.2 and 7.8, ArCHHCH), 3.11 (3H, s, OMe), 3.17 (1H, dd, J 14.2 and 3.3, ArCHHCH), 3.34 (3H, s, NMe), 3.35 (1H, m, OCHHCH<sub>2</sub>OMe), 3.47 (2H, m, OCH<sub>2</sub>CH<sub>2</sub>OMe), 3.55 (1H, m, OCHHCH<sub>2</sub>OMe), 3.70 (2H, m, CH<sub>2</sub>OH), 3.94 (2H, t, J 5.3, NCH<sub>2</sub>), 3.95 (1H, dd, J 7.8 and 3.3, ArCH<sub>2</sub>CH), 4.26 (2H, t, J 5.3, NCH<sub>2</sub>CH<sub>2</sub>O), 4.96 (1H, m, PhCH), 6.83 (2H, d, J 8.7, aryl 3-H and 5-H), 6.95–7.35 (11H, m, aryl-H) and 7.57 (1H, br, exchanges with D<sub>2</sub>O, NH).

4.8.6. [2S,N(1S)]-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2-methoxyethoxy)-N-(2-hydroxy-1-phenylethyl)propanamide **7c** 

Similarly prepared. A gum, 27%; (found M<sup>+</sup> (EI) 533.2526.  $C_{30}H_{35}N_3O_6$  requires M 533.2526);  $[\alpha]_D^{25}$  –33 (c 1.1 in CHCl<sub>3</sub>); d.e. 92.6% by HPLC;  $\delta_H$  (400 MHz, CDCl<sub>3</sub>) 2.81 (1H, dd, J 14.3 and 8.3, ArCHHCH), 3.07 (1H, dd, J 14.3 and 3.2, ArCHHCH), 3.30 (1H, br, exchanges with D<sub>2</sub>O, OH), 3.35 (3H, s, OMe), 3.36 (3H, s, NMe), 3.48–3.65 (4H, m, OCH<sub>2</sub>CH<sub>2</sub>OMe), 3.71 (1H, dd, J 11.8 and 7.5, NCHPhCHHOH), 3.82 (1H, dd, J 11.8 and 4.0, NCHPhCHHOH), 3.93 (2H, t, J 5.3, NCH<sub>2</sub>), 3.94 (1H, dd, J 8.3 and 3.2, ArCH<sub>2</sub>CH), 4.22 (2H, t, J 5.3, NCH<sub>2</sub>CH<sub>2</sub>O), 5.05 (1H, m, PhCH), 6.85 (2H, d, J 8.7, aryl 3-H and 5-H), 6.95–7.35 (11H, m, aryl-H) and 7.54 (1H, br, exchanges with D<sub>2</sub>O, NH).

4.9. General procedure for reaction of racemic **1d** and **1e** with (S)-4-benzyloxazolidin-2-one. Preparation of diastereoisomeric 3-(trifluoromethyl)phenoxyacyloxazolidin-2-ones **11e** and **12e** 

Oxalyl chloride (2.62 mL) was added dropwise to a solution of the acid **1e** (3.0 g, 6.0 mmol) in dry benzene (120 mL). The mixture was heated at reflux for 2.5 h, cooled and evaporated in vacuo. Repeated re-evaporation from dry THF afforded the crude acid chloride, a gum.

A solution of *n*-butyllithium (1.6 M in hexane, 7.5 mL, 12.0 mmol) was added dropwise, over 10 min, to a -70°C solution of (*S*)-4-benzyloxazolidin-2-one (2.12 g, 6.0 mmol) in dry THF (50 mL) under

argon. The mixture was stirred at  $-70^{\circ}$ C for an additional 10 min prior to the addition, over 10 min, of a solution of the acid chloride (above) in dry THF (100 mL). Stirring was continued at  $-70^{\circ}$ C for 1 h and the mixture was then allowed to warm to room temperature overnight (20 h), then concentrated. The residue was diluted with water (500 mL), extracted with ethyl acetate (3×500 mL) and the combined ethyl acetate solutions washed with water (500 mL) and brine (500 mL), dried and evaporated. Chromatography using ethyl acetate:isohexane (30:70) as eluent afforded firstly the [3(2R),4R]-diastereoisomer **11e**, followed by the [3(2R),4R]-diastereoisomer **12e**.

# 4.9.1. [3(2R),4S]-3-[3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-[3-(trifluoro-methyl)phenoxy]propanoyl]-4-benzyloxazolidin-2-one 11e

A white foam (1.08 g, 28%); (found (M+H)<sup>+</sup> (CI) 660.2319.  $C_{36}H_{32}F_{3}N_{3}O_{6}$  requires M 660.2322);  $[\alpha]_{D}^{25}$  +29 (c 1.00 in CHCl<sub>3</sub>); d.e. 100% by HPLC;  $\nu_{max}$  (KBr)/cm<sup>-1</sup> 1780, 1710 and 1645 (CO);  $\delta_{H}$  (400 MHz, CDCl<sub>3</sub>) 2.72 (1H, dd, J 13.5 and 9.6, PhCHHCH), 3.11 (1H, dd, J 14.0 and 8.9, ArCHHCH), 3.23 (1H, dd, J 13.5 and 3.3, PhCHHCH), 3.26 (1H, dd, J 14.0 and 3.2, ArCHHCH), 3.32 (3H, s, NMe), 3.92 (2H, t, J 5.3, NCH<sub>2</sub>), 4.23 (2H, t, J 5.3, OCH<sub>2</sub>CH<sub>2</sub>N), 4.25 (1H, dd, J 9.0 and 3.5, oxazolidinone 5a-H), 4.31 (1H, t, J 9.0, oxazolidinone 5b-H), 4.71 (1H, m, oxazolidinone 4-H), 5.97 (1H, dd, J 8.9 and 3.3, ArCH<sub>2</sub>CH), 6.83 (2H, d, J 8.7, aryl 3-H and 5-H), 6.92 (1H, dd, J 8.3 and 2.6, CF<sub>3</sub>-phenoxy 6-H) and 6.99–7.36 (14H, m, aryl-H); m/z (CI) 660 [(M+H)<sup>+</sup>, 100%], 500 (9), 483 (4), 371 (5), 325 (7), 297 (7) and 195 (85).

# 4.9.2. [3(2S),4S]-3-[3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-[3-(trifluoro-methyl)phenoxy]propanoyl]-4-benzyloxazolidin-2-one **12e**

A white foam (0.73 g, 19%); (found (M+H)<sup>+</sup> (CI) 660.2319.  $C_{36}H_{32}F_{3}N_{3}O_{6}$  requires M 660.2322);  $[\alpha]_{D}^{25}$  +62 (c 1.01 in CHCl<sub>3</sub>); d.e. 97.2% by HPLC;  $\nu_{max}$  (KBr)/cm<sup>-1</sup> 1780, 1710 and 1640 (CO);  $\delta_{H}$  (400 MHz, CDCl<sub>3</sub>) 2.73 (1H, dd, J 13.5 and 9.5, PhCHHCH), 3.18 (2H, d, J 6.6, ArC $H_{2}$ CH), 3.18 (1H, dd, J 13.5 and 3.3, PhCHHCH), 3.32 (3H, s, NMe), 3.92 (2H, t, J 5.3, NCH<sub>2</sub>), 4.02 (1H, dd, J 9.1 and 7.7, oxazolidinone 5a-H), 4.15 (1H, dd, J 9.1 and 2.4, oxazolidinone 5b-H), 4.23 (2H, t, J 5.3, OC $H_{2}$ CH<sub>2</sub>N), 4.52 (1H, m, oxazolidinone 4-H), 6.13 (1H, t, J 6.6, ArCH<sub>2</sub>CH), 6.82 (2H, d, J 8.7, aryl 3-H and 5-H) and 6.95–7.40 (15H, m, aryl-H); m/z (CI) 660 [(M+H)<sup>+</sup>, 100%], 500 (15), 297 (4) and 195 (92).

# 4.9.3. [3(2R),4S]-3-[3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-benzyloxy-propanoyl]-4-benzyloxazolidin-2-one 11d

Similarly prepared. A foam, 24%; (found M<sup>+</sup> (EI) 605.2525.  $C_{36}H_{35}N_3O_6$  requires M 605.2526);  $[\alpha]_D^{25}$  +32 (c 1.09 in CHCl<sub>3</sub>); d.e. 97.8% by HPLC;  $\delta_H$  (270 MHz, CDCl<sub>3</sub>) 2.66 (1H, dd, J 13.2 and 9.3, PhCHHCH), 2.80–3.20 (3H, m, ArC $H_2$  and PhCHHCH), 3.34 (3H, s, NMe), 3.94 (2H, t, J 5.5, NCH<sub>2</sub>), 4.10–4.30 (4H, m, OC $H_2$ CH<sub>2</sub>N and oxazolinone 5-H<sub>2</sub>), 4.38 (1H, d, J 11.6, PhCHHO), 4.51 (1H, d, J 11.6, PhCHHO), 4.55 (1H, m, oxazolidinone 4-H), 5.21 (1H, dd, J 9.3 and 3.8, ArCH<sub>2</sub>CH), 6.81 (2H, d, J 8.8, aryl 3-H and 5-H) and 6.95–7.40 (16H, m, aryl-H).

# 4.9.4. [3(2S),4S]-3-[3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-benzyloxy-propanoyl]-4-benzyloxazolidin-2-one **12d**

Similarly prepared. A gum, 17%; (found M<sup>+</sup> (EI) 605.2525.  $C_{36}H_{35}N_3O_6$  requires M 605.2526);  $[\alpha]_D^{25}$  +39 (c 1.02 in CHCl<sub>3</sub>); d.e. 89.8% by HPLC;  $\delta_H$  (400 MHz, CDCl<sub>3</sub>) 2.66 (1H, dd, J 13.5 and 9.5, PhCHHCH), 2.92 (1H, dd, J 13.6 and 8.3, ArCHHCH), 3.00 (1H, dd, J 13.6 and 4.8, ArCHHCH), 3.17 (1H, dd, J 13.5 and 3.3, PhCHHCH), 3.33 (3H, s, NMe), 3.93 (2H, t, J 5.3, NCH<sub>2</sub>), 4.04 (1H, m, oxazolidinone 5a-H), 4.10 (1H, m, oxazolidinone 5b-H), 4.24 (2H, t, J 5.3, OCH<sub>2</sub>CH<sub>2</sub>N), 4.43 (1H, d, J

- 11.7, PhCHHO), 4.51 (1H, m, oxazolidinone 4-H), 4.53 (1H, d, *J* 11.7, PhCHHO), 5.33 (1H, dd, *J* 8.3 and 4.8, ArCH<sub>2</sub>CH), 6.80 (2H, d, *J* 8.6, aryl 3-H and 5-H) and 6.95–7.40 (16H, m, aryl-H).
- 4.10. General procedure for hydrolysis of diastereoisomeric amides 6 and 7
- 4.10.1. (R)-(+)-3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl-2-ethoxypropanoic acid 8a A solution of amide 6a (1.137 g, 2.3 mmol), sulphuric acid (1.0 M, 23 mL), water (46 mL) and dioxan (46 mL) was heated at 90°C for 48 h, cooled, diluted with water (300 mL) and the pH adjusted to 2.5 by addition of saturated aqueous sodium bicarbonate solution. The mixture was extracted with ethyl acetate (3×250 mL) and the combined organic solutions washed with water (500 mL) and brine (500 mL), dried and evaporated. The residue was chromatographed with methanol:dichloromethane (gradient, 1.5:98.5 to 3.5:96.5) as eluent to afford acid 8a, (0.324 g, 37%), a foam, spectroscopically identical with racemate 1a; (found M<sup>+</sup> (EI) 384.1687. C<sub>21</sub>H<sub>24</sub>N<sub>2</sub>O<sub>5</sub> requires *M* 384.1685); [α]<sub>D</sub><sup>25</sup> +15 (*c* 0.90 in MeOH); e.e. 93.2% by HPLC.
- 4.10.2. (S)-(-)-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl-2-ethoxypropanoic acid **9a** Similarly prepared. A foam, 38%; (found M<sup>+</sup> (EI) 384.1684.  $C_{21}H_{24}N_2O_5$  requires *M* 384.1685);  $[\alpha]_D^{25}$  -16 (*c* 1.08 in MeOH); e.e. 93.6% by HPLC; otherwise identical with racemate **1a**.
- 4.10.3. (R)-(+)-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl-2-(2,2,2-trifluoroethoxy)-propanoic acid  ${\it 8b}$

Similarly prepared. A solid, 51%; m.p. 115°C; (found C, 57.8; H, 4.7; N, 6.7%; M<sup>+</sup> (EI) 438.1403.  $C_{21}H_{21}F_3N_2O_5$  requires C, 57.5; H, 4.8; N, 6.4%; *M* 438.1403);  $[\alpha]_D^{25}$  +24 (*c* 0.31 in CHCl<sub>3</sub>); e.e. 96.6% by HPLC; otherwise identical with racemate **1b**.

4.10.4. (S)-(-)-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2,2,2-trifluoroethoxy)-propanoic acid  $\bf 9b$ 

Similarly prepared. A solid, 48%; m.p. 116–117°C; (found C, 57.9; H, 4.7; N, 6.8%; M<sup>+</sup> (EI) 438.1403.  $C_{21}H_{21}F_3N_2O_5$  requires C, 57.5; H, 4.8; N, 6.4%; *M* 438.1403); [ $\alpha$ ]<sub>D</sub><sup>25</sup> –25 (c 0.24 in CHCl<sub>3</sub>); e.e. 95.2% by HPLC; otherwise identical with racemate **1b**.

 $4.10.5. \ \ (R)-(+)-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy] phenyl]-2-(2-methoxyethoxy) propanoic\ acid\ \textbf{8c}$ 

Similarly prepared. A foam, 60%; (found M<sup>+</sup> (EI) 414.1791.  $C_{22}H_{26}N_2O_6$  requires *M* 414.1791);  $[\alpha]_D^{25}$  +28 (*c* 0.63 in CHCl<sub>3</sub>); e.e. 94% by HPLC; otherwise identical with racemate **1c**.

4.10.6. (S)-(-)-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-(2-methoxyethoxy)propanoic acid **9c** 

Similarly prepared. A foam, 43%; e.e. 88% by HPLC. Repeated crystallisation of the (*S*)-(–)-1-phenylethylamine salt subsequently afforded acid 9c, a foam; (found C, 63.6; H, 6.4; N, 6.8%;  $M^+$  (EI) 414.1791.  $C_{22}H_{26}N_2O_6$  requires C, 63.8; H, 6.3; N, 6.8%; M 414.1791);  $[\alpha]_D^{25}$  –28 (c 0.63 in CHCl<sub>3</sub>); e.e. 94% by HPLC; otherwise identical with racemate 1c.

4.11. General procedure for recovery of chiral  $\alpha$ -oxyacids 8d,e and 9d,e from diastereoisomeric acyloxazolidin-2-ones 11 and 12

4.11.1. (R)-(+)-3-[4-[2-[N-(2-benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-[3-(trifluoromethyl)-phenoxy]propanoic acid 8e

A freshly prepared solution of sodium methoxide in methanol [sodium hydride (60% dispersion in oil, 0.064 g, 1.6 mmol) dissolved in methanol (2.5 mL)] was added to a 0°C solution of imide **11e** (0.940 g, 1.4 mmol) in methanol (30 mL). The mixture was stirred at 0°C for 13 min, then quenched by the addition of dilute hydrochloric acid (2.0 M, 0.7 mL, 1.4 mmol) and concentrated in vacuo. The residue was suspended in water (100 mL) and extracted with ethyl acetate (3×200 mL). The combined ethyl acetate solutions were washed with brine, dried and evaporated. Chromatography using ethyl acetate:dichloromethane (3:97) as eluent afforded the ester **13e** (0.586 g, 81%), a gum, spectroscopically identical with racemate **5e**; (found M<sup>+</sup> (EI) 514.1717.  $C_{27}H_{25}F_3N_2O_5$  requires *M* 514.1716);  $[\alpha]_D^{25}$  +12 (*c* 1.33 in CHCl<sub>3</sub>); e.e. 97% by HPLC.

A solution of ester 13e (0.543 g, 1.1 mmol), dilute hydrochloric acid (2.0 M, 55 mL) and dioxan (55 mL) was heated at 90°C for 6.5 h then concentrated in vacuo. The residue was suspended in brine (150 mL) and sufficient aqueous sodium hydroxide solution (2.5 M) was added until the solution was at pH 1.5. The mixture was extracted with ethyl acetate (3×300 mL) and the combined ethyl acetate solutions dried and evaporated to afford the crude acid 8e, a solid (0.511 g, 93%), spectroscopically identical with racemate 1e. A sample of this material was recrystallised form dichloromethane–hexane to afford an analytical sample, m.p. 151–152°C; (found C, 62.0; H, 4.5; N, 5.4%; M<sup>+</sup> (EI) 500.1568.  $C_{26}H_{23}F_3N_2O_5$  requires C, 62.4; H, 4.6; N, 5.6%; M 500.1559); [ $\alpha$ ]<sub>D</sub><sup>25</sup> +8 (c 0.7 in MeOH); e.e. 98% by HPLC.

4.11.2. (S)-(-)-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-[3-(trifluoromethyl)-phenoxy]propanoic acid  $\mathbf{9e}$ 

Similarly prepared. Methyl ester **14e**: a gum, 86%;  $[\alpha]_D^{25}$  –12 (c 1.2 in CHCl<sub>3</sub>); e.e. 96.4% by HPLC. Acid **9e**: a solid, 85%; m.p. 149–151°C; (found C, 62.3; H, 4.6; N, 5.5%.  $C_{26}H_{23}F_3N_2O_5$  requires C, 62.4; H, 4.6; N, 5.6);  $[\alpha]_D^{25}$  –7 (c 1.09 in MeOH); e.e. 98% by HPLC; otherwise identical with racemate **1e**.

4.11.3. (R)-(+)-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-benzyloxypropanoic acid  ${\it 8d}$ 

Similarly prepared. Methyl ester **13d**: a gum, 80%; (found M<sup>+</sup> (EI) 460.2000.  $C_{27}H_{28}N_2O_5$  requires M 460.1999);  $[\alpha]_D^{25}$  +31 (c 0.4 in CHCl<sub>3</sub>); e.e. 96% by HPLC. Acid **8d**: a foam, 68%;  $[\alpha]_D^{25}$  +24 (c 1.00 in CHCl<sub>3</sub>); e.e. 96% by HPLC; otherwise identical with racemate **1d**.

4.11.4. (S)-(-)-3-[4-[2-[N-(2-Benzoxazolyl)-N-methylamino]ethoxy]phenyl]-2-benzyloxypropanoic acid  $\bf{9d}$ 

Methyl ester **14d**: a gum, 75%;  $[\alpha]_D^{25}$  –34 (*c* 0.55 in CHCl<sub>3</sub>); e.e. 88% by HPLC. Acid **9d**: a foam, 71%;  $[\alpha]_D^{25}$  –27 (*c* 1.46 in CHCl<sub>3</sub>); e.e. 88% by HPLC; otherwise identical with racemate **1d**.

### References

1. Previous paper in this series: Haigh, D.; Allen, G.; Birrell, H. C.; Buckle, D. R.; Cantello, B. C. C.; Eggleston, D. S.; Haltiwanger, R. C.; Holder, J. C.; Lister, C. A.; Pinto, I. L.; Rami, H. K.; Sime, J. T.; Smith, S. A.; Sweeney, J. D. *Bioorg. Med. Chem.* 1999, 7, 821.

- 2. (a) Reaven, G. M. Diabetes 1988, 37, 1595. (b) Kahn, C. R. Diabetes 1994, 43, 1066.
- 3. Bjorntorp, P.; Fahlen, M.; Grimby, G.; Gustafson, A.; Holm, J.; Renstrom, P.; Schersten, T. Metabolism 1972, 21, 1037.
- (a) Larson, E. R.; Clark, D. A.; Stevenson, R. W. Ann. Reports Med. Chem. 1989, 25, 205. (b) Colca, J. R.; Tanis, S. P. Ann. Reports Med. Chem. 1992, 27, 219. (c) Colca, J. R.; Morton, D. R. In New Antidiabetic Drugs; Bailey, C. J.; Flatt, P. R. Eds. Smith-Gordon, 1990; p. 255. (d) Dow, R. L.; Kreutter, D. K. Ann. Reports Med. Chem. 1995, 30, 159. (e) Hulin, B.; McCarthy, P. A.; Gibbs, E. M. Current Pharmaceutical Design 1996, 2, 85.
- 5. Yoshioka, T.; Fujita, T.; Kanai, T.; Aizawa, Y.; Kurumada, T.; Hasegawa, K.; Horikoshi, H. J. Med. Chem. 1989, 32, 421.
- 6. Cantello, B. C. C.; Cawthorne, M. A.; Cottam, G. P.; Duff, P. T.; Haigh, D.; Hindley, R. M.; Lister, C. A.; Smith, S. A.; Thurlby, P. L. *J. Med. Chem.* **1994**, *37*, 3977.
- (a) Lehmann, J. M.; Moore, L. B.; Smith-Oliver, T. A.; Wilkison, W. O.; Willson, T. M.; Kliewer, S. A. J. Biol. Chem. 1995, 270, 12953–12956. (b) Forman, B. M.; Tontonoz, P.; Chen, J.; Brun, R. P.; Spiegelman, B. M.; Evans, R. M. Cell 1995, 83, 803. (c) Willson, T. M.; Cobb, J. E.; Cowan, D. J.; Wiethe, R. W.; Correa, I. D.; Prakash, S. R.; Beck, K. D.; Moore, L. B.; Kliewer, S. A.; Lehmann, J. M. J. Med. Chem. 1996, 39, 665. (d) Young, P. W.; Buckle, D. R.; Cantello, B. C. C.; Chapman, H.; Clapham, J. C.; Coyle, P. J.; Haigh, D.; Hindley, R. M.; Holder, J. C.; Kallender, H.; Latter, A. L.; Laurie, K. W. W.; Mossakowska, D.; Murphy, G. J.; Cox, L. R.; Smith, S. A. J. Pharmacol. Exp. Ther. 1998, 284, 751.
- 8. Buckle, D. R.; Cantello, B. C. C.; Cawthorne, M. A.; Dean, D. K.; Faller, A.; Haigh, D.; Hindley, R. M.; Jefcott, L. J.; Lister, C. A.; Pinto, I. L.; Rami, H. K.; Smith, D. G.; Smith, S. A. Bioorg. Med. Chem. Lett. 1996, 6, 2121.
- 9. Thiazolidinediones are known to undergo rapid racemisation under physiological conditions. Sohda, T.; Mizuno, K.; Kawamatsu, Y. *Chem. Pharm. Bull.* **1984**, *32*, 4460.
- (a) Haigh, D.; Sime, J. T. PCT International Patent Appl., Publication No. WO 9401420, 1994.
   (b) Miller, D. J.; Moody, C. J. Tetrahedron 1995, 51, 10811.
- 11. Compounds **3a** and **3e** are known compounds. See (a) Grell, W.; Machleidt, H. *Liebigs Ann. Chem.* **1966**, 699, 53, and (b) Haigh, D. *Tetrahedron* **1994**, 50, 3177, respectively.
- 12. Olefin geometry was determined by comparison of the gated decoupled <sup>13</sup>C NMR spectrum <sup>3</sup>J<sub>HC</sub> coupling constant between the olefinic proton and the ester carbonyl carbon atom with literature values. See, for example, (a) Hansen, P. E. *Prog. Nucl. Magn. Reson. Spectrosc.* **1981**, *14*, 175. (b) Vogeli, U.; Von Philipsborn, W. *Org. Magn. Reson.* **1975**, 7, 617. (c) Kingsbury, C. A.; Draney, D.; Sopchik, A.; Rissler, W.; Durham, D. *J. Org. Chem.* **1976**, *41*, 3863. (d) Meanwell, N. A.; Roth, H. R.; Smith, E. C. R.; Wedding, D. L.; Wright, J. J. K. *J. Org. Chem.* **1991**, *56*, 6897.
- For alternative syntheses of related compounds see, for example, (a) Horner, L.; Renth, E.-O. Liebigs Ann. Chem. 1967, 703, 37. (b) Cacchi, S.; Ciattini, P. G.; Morera, E.; Ortar, G. Tetrahedron Lett. 1987, 28, 3039. (c) Sakamoto, T.; Kondo, Y.; Kashiwagi, Y.; Yamanaka, H. Heterocycles 1988, 27, 257. (d) Aitken, R. A.; Thom, G. L. Synthesis 1989, 958. (e) Merlic, C. A.; Semmelhack, M. F. J. Organomet. Chem. 1990, 391, C23.
- 14. Youn, K.; Yon, G. H.; Pak, C. S. Tetrahedron Lett. 1986, 27, 2409.
- 15. Jacques, J.; Collet, A.; Wilen, S. H. Enantiomers, Racemates and Resolutions; Wiley, 1981.
- 16. Resolution of related thiazolidinedione analogues was only achieved by formation of the quinine salt. See footnote in Cantello, B. C. C.; Eggleston, D. S.; Haigh, D.; Haltiwanger, R. C.; Heath, C. M.; Hindley, R. M.; Jennings, K. R.; Sime, J. T.; Woroniecki, S. R. J. Chem. Soc., Perkin Trans. 1 1994, 3319.
- 17. (a) Kagan, H. B.; Fiaud, J. C. *Topics in Stereochemistry* **1988**, *18*, 249. (b) Chinchilla, R.; Najera, C.; Yus, M.; Heumann, A. *Tetrahedron: Asymmetry* **1991**, 2, 101. (c) Mazon, A.; Najera, C.; Yus, M.; Heumann, A. *Tetrahedron: Asymmetry* **1992**, 3, 1455.
- 18. Ward, R. S. Tetrahedron: Asymmetry 1995, 6, 1475.
- 19. Pirkle, W. H.; Finn, J. In Asymmetric Synthesis; Morrison, J. D. Ed. Academic Press, 1983; vol. 1, p. 87.
- See, for example, (a) Eisenbraun, E. J.; Adolphen, G. H.; Schorno, K. S.; Morris, R. N. J. Org. Chem. 1971, 36, 414. (b) Gossauer, A.; Weller, J.-P. J. Am. Chem. Soc. 1978, 100, 5928. (c) Franck, A.; Ruchardt, C. Chem. Lett. 1984, 1431. (d) Larsen, R. D.; Corley, E. G.; Davis, P.; Reider, P. J.; Grabowski, E. J. J. J. Am. Chem. Soc. 1989, 111, 7650. (e) Rettinger, K.; Burschka, C.; Scheeben, P.; Fuchs, H.; Mosandl, A. Tetrahedron: Asymmetry 1991, 2, 965.
- (a) Helmchen, G.; Ott, R.; Sauber, K. *Tetrahedron Lett.* 1972, 3873. (b) Helmchen, G. *Tetrahedron Lett.* 1974, 1527. (c) Helmchen, G.; Strubert, W. *Chromatographia* 1974, 7, 713. (d) Helmchen, G.; Volter, H.; Schule, W. *Tetrahedron Lett.* 1977, 1417. (e) Numan, H.; Wynberg, H. *J. Org. Chem.* 1978, 43, 2232. (f) Oi, S.; Matsuzaka, Y.; Yamashita, J.; Miyano, S. *Bull. Chem. Soc. Jpn.* 1989, 62, 956.
- 22. (a) Helmchen, G.; Nill, G.; Flockerzi, D.; Schule, W.; Youssef, M. S. K. Angew. Chem., Int. Ed. Engl. 1979, 18, 62. (b) Helmchen, G.; Nill, G.; Flockerzi, D.; Youssef, M. S. K. Angew. Chem., Int. Ed. Engl. 1979, 18, 63. (c) Ade, E.; Helmchen, G.; Heiligenmann, G. Tetrahedron Lett. 1980, 21, 1137. (d) Karl, V.; Kaunzinger, A.; Gutser, J.; Steuer, P.; Angles-Angel, J.; Mosandl, A. Chirality 1994, 6, 420.

- (a) Song, C. E.; Lee, S. G.; Lee, K. C.; Kim, I. O. *J. Chromatogr.* 1993, 654, 303. (b) Li, G.; Patel, D.; Hruby, V. J. *J. Chem. Soc., Perkin Trans. I* 1994, 3057. (c) Koll, P.; Lutzen, A. *Tetrahedron: Asymmetry* 1995, 6, 43. (d) Xiang, L.; Wu, H.; Hruby, V. J. *Tetrahedron: Asymmetry* 1995, 6, 83. (e) Burke Jr, T. R.; Barchi Jr, J. J.; George, C.; Wolf, G.; Shoelson, S. E.; Yan, X. *J. Med. Chem.* 1995, 38, 1386.
- 24. Fuhry, M. A. M.; Holmes, A. B.; Marshall, D. R. J. Chem. Soc., Perkin Trans. 1 1993, 2743.
- 25. (a) Evans, D. A.; Bartroli, J.; Shih, T. L. *J. Am. Chem. Soc.* **1981**, *103*, 2127. (b) Faull, A. W.; Brewster, A. G.; Brown, G. R.; Smithers, M. J.; Jackson, R. *J. Med. Chem.* **1995**, *38*, 686.
- 26. See, for example, (a) Yamaguchi, S. In *Asymmetric Synthesis*; Morrison, J. D. Ed. Academic Press, 1983; vol. 1, p. 125. (b) Trost, B. M.; Bunt, R. C.; Pulley, S. R. *J. Org. Chem.* **1994**, *59*, 4202.
- 27. Birrell, H. C.; Cantello, B. C. C.; Eggleston, D. S.; Haigh, D.; Haltiwanger, R. C.; Hindley, R. M.; Ramaswamy, A.; Stevens, N. C. *Tetrahedron: Asymmetry* **1999**, *10*, 1353.